

PRINCE OF SONGKLA UNIVERSITY
FACULTY OF ENGINEERING

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| Final Examination : Semester II | Academic Year : | 2002 |
| Date : 20 February 2003 | Time : | 9.00-12.00 |
| Subject : 230-630 Advanced Transport Phenomena I | Room : | R300 |

- ข้อสอบมี 5 ข้อ ต้องทำทุกข้อ คะแนนเต็ม 80 คะแนน
- ควรใช้เวลาทำข้อสอบโดยเฉลี่ย 2 นาที/คะแนน
- อนุญาตให้นำหนังสือ เอกสาร เครื่องคำนวณ และอุปกรณ์อื่น ๆ เข้าห้องสอบได้

สุธรรม สุขมณี

ผู้ออกข้อสอบ

9 กุมภาพันธ์ 2546

- 1) Predict the mass diffusivity of the gas mixture with 70 mole percent of Ethylene (C_2H_4) and 30 mole percent of Ethane (C_2H_6) at 347.4 K and 29.7 atm. (15 points)

- 2) Consider a ternary system consists of species A (Molecular weight of 28, Concentration of 0.04 mole/cm³), B (Molecular weight of 2, Concentration of 0.06 mole/cm³) and AB (Molecular weight of 30, Concentration of 0.05 mole/cm³) at given time and position. If the absolute velocities of A and B relative to stationary coordinates (v_A and v_B) are +6 and +8 cm/s respectively and the local molar average velocity of the gas mixture is +2.8 cm/s. Determine the following quantities :
 - 2.1 Local absolute velocities of the species AB .
 - 2.2 Local mass average velocity of the gas mixture.
 - 2.3 Molar flux of B relatives to stationay coordinates.
 - 2.4 Mass flux of A relatives to mass average velocity.
 - 2.5 Molar flux of AB relatives to molar average velocity.

(10 points)

- 3) Derive an expression for the steady concentration distribution (C_A) and mass transfer rate (W_A) of gas A in the hollow porous sphere with a constant effective mass diffusivity of D_A , an inside radius of λR and an outside radius of R . The concentration of A inside and outside the sphere are $C_{A\lambda}$ and C_{AR} respectively. For the diffusion of gas A in porous medium at constant temperature and pressure, one may assume that the molar flux of A relative to stationary plane as:

$$N_A = -D_A (dC_A/dr)$$

(20 points)

- 4) Dry air with an uniform temperature of 30 °C and a pressure of 1.46 atm. ($\rho = 1.7 \text{ kg/m}^3$, $\mu = 0.01822 \text{ mPa}\cdot\text{s}$) is flowing in a smooth circular pipe of diameter 54.1 mm with a mass flow rate of 69.7 kg/h (Reynolds number of about 25000 and the wall shear stress (τ_w) for the flowing air stream may be taken as 0.13 N/m²). Beginning at $z = 0$ to $z = 2000$ mm, there is a diffusing device that transfer water vapor from the tube wall to an air stream with a constant molar flux (N_{A0}) of $-0.018 \text{ mole/m}^2\cdot\text{s}$. At the distance of 2000 mm from the start of this constant wall molar flux, the time-smoothed concentration of water vapor (\bar{C}_A) at the pipe center-line is 0.8 mole/m³ and the concentration of water vapor at the pipe wall (C_{A0}) is 2 mole/m³. The diffusivity of water vapor in air (D_{AB}) at 30 °C is $1.78 \times 10^{-5} \text{ m}^2/\text{s}$. Find the time-smoothed concentration of water vapor (\bar{C}_A) and the turbulent diffusivity ($\bar{D}_{AB}^{(t)}$) at the distance of 13.525 mm from the pipe wall.

(20 points)

Assumption In turbulent core and low molar flux : $\bar{J}_A^{(t)} \cong N_{A0} \cong -\bar{D}_{AB}^{(t)} \frac{d\bar{C}_A}{dr}$

- 5) Consider a cylindrical rod 2 inches in diameter of Naphthalene (Molecular weight of 128.2) placed in dry air at 115 °F (Density of 0.0692 lb/ft³ and Viscosity of 0.0469 lb/ft-h) flowing perpendicular to the axis with a velocity of 15 ft/s. The vapor pressure of solid Naphthalene is 0.565 mmHg and the diffusivity of Naphthalene in air at 115 °F is $7.5 \times 10^{-5} \text{ ft}^2/\text{s}$. Determine the mass transfer coefficient and the sublimation rate of Naphthalene to air per unit length of the rod at 115 °F and 760 mmHg. (15 points)