

PRINCE OF SONGKLA UNIVERSITY
FACULTY OF ENGINEERING

Final Examination : Semester II
Date : 18 February 2004
Subject : 230-432 Chemical Engineering
Plant Design

Academic Year : 2003
Time : 13.30-16.30
Room : R 300

Student Name: ID No :

Number of questions : 4

Time : 3 hours

Total marks : 100

Lecture notes are allowed

Calculators are allowed

Question	Full Marks	Marks Received
1	25	
2	25	
3	25	
4	25	
Total	100	

Question 1:

- a) Two mutually exclusive projects have the following cash flows.

Year	Project I	Project II
0	-\$100,000	-\$60,000
1	\$60,000	\$36,500
2	\$60,000	\$36,500

Which of these two projects should the company choose if the interest rate is 10% per year?

(10 marks)

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- b) A pulp and paper company has two alternatives in buying an evaporator. Two types of evaporator have different installed and maintenance costs, salvage values and service lives, as shown below.

	Evaporator A	Evaporator B
Installed cost	\$18,000	\$25,000
Annual maintenance cost	\$4,000	\$3,000
Salvage value	\$500	\$1,500
Service life	2 year	3 year

Use the interest rate 10% to select an evaporator.

(15 marks)

Answer Q1

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Answer Q1 (continued)

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Question 2:

A chemical plant is proposed with \$90 millions fixed capital investment which is equally distributed during the 3-year construction period. The plant requires \$40 millions of working capital at the beginning of first-year production. At 100% capacity the plant has sales revenues of \$150 millions/year and the total cost of product excluding depreciation of \$100 millions/year. The plant is projected to operate at 50% and 75% capacity during the first and second operating years. The number of operators is not reduced proportionally therefore the total cost of product is projected to reduce by 45% and 32% during the first and second operating years respectively. The depreciable capital depreciates in 6 years after beginning of operation with values 18.0, 28.8, 17.28, 10.37, 10.37 and 5.18 \$millions. Tax rate is 37%.

Write annual cash flow for 15-year project life. Calculate the net present value for an interest rate of 15% and the internal rate of return.

(25 marks)

Answer Q2

Student Name : ID No :

Answer Q2 (continued)

Student Name : ID No :

Question 3:

- a) A sieve-tray distillation column for separation of isobutene and n-butane mixture has flows and property profiles on tray number 4 as shown below. Assume that tray spacing is 24 inches, foaming factor is 1.0 and $A_h/A_a > 0.1$. Estimate the column diameter at this tray location.

Conditions at tray number 4:

	Liquid	Vapour
Mass flow (lb/hr)	215,000	244,000
Molecular Weight	58.12	58.12
Temperature (°F)	130	132
Density (lb/ft ³)	32.4	1.095
Surface tension (dyne/cm)	7.1	----

Note: 1 ft = 0.3048 m

(15 marks)

- b) If this tray column is made of carbon steel, 212 feet high and operates at a pressure of 110 psia or 6.75 bar(guage), estimate the cost of the column (without trays), C_{Notray} .

$$C_{\text{Notray}} = 1780L^{0.87}D^{1.23}[2.86+1.694F_M(10.01-7.408\ln P+1.395(\ln P)^2)]$$

where C_{Notray} = cost of the column (without trays) in \$

L = column height in meters

D = column diameter in meters

F_M = material factor = 1.0 for carbon steel

P = design pressure in bar (guage)

(5 marks)

- c) Specify parameters for choosing between a tray column and a packed column.

(5marks)

Student Name : ID No :

Answer Q3

Student Name : ID No :

Answer Q3 (continued)

Student Name : ID No :

Question 4:

- a) Explain the following terms in sieve tray design.
- a.1 Trial tray lay-out
 - a.2 Weeping
 - a.3 Entrainment
 - a.4 Turndown ratio
- (10 marks)
- b) Discuss the use of “Glass lining” in chemical processing units. What are the applications and limitations of this technique?
- (3 marks)
- c) What is dust explosion? How does it occur? How to prevent it?
- (3 marks)
- d) Describe the method for determination of optimum conditions such as minimum total cost when there are two independent variables x and y.
- (5 marks)
- e) For inherent safety plant design, discuss the “Attenuation” approach.
- (4 marks)

----- END of question

Answer Q4

Student Name : ID No :

Answer Q4(continued)

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