

Name: \_\_\_\_\_ Student ID \_\_\_\_\_  
Nickname: \_\_\_\_\_ Group: \_\_\_\_\_

**Prince of Songkla University**  
**Faculty of Engineering**

**Exam: Mid-Term, Semester I**  
**Date: August 5, 2004**  
**Subject: 230-391**  
**Basic Chemical Engineering I**

**Academic Year: 2004 – 2005**  
**Time: 1:30 – 4:30 PM**  
**Room: R300**

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ทฤษฎีในการสอบโทษขั้นต่ำคือ ปรับตกในรายวิชาที่ทฤษฎี และพักการเรียน 1 ภาคการศึกษา

**Instructions:** There are a total of 5 problems and 9 pages (not including this page). Place your name and the student ID number on every page. Students are allowed to use only a pen or pencil and a calculator. They can also bring in 1 sheet of A4 front side only, a Conversions Table, and a Dictionary. No exams are allowed to leave the room.

<b>Points Distribution (For Grader Only)</b>		
<b>Problem</b>	<b>Points Value</b>	<b>Score</b>
<b>1</b>	<b>15</b>	
<b>2</b>	<b>20</b>	
<b>3</b>	<b>20</b>	
<b>4</b>	<b>20</b>	
<b>5</b>	<b>25</b>	
<b>Total</b>	<b>100</b>	

**Exam prepared by**  
**Ram Yamsaengsung**  
**July 27, 2004**

**PLEASE CHECK TO MAKE SURE THAT**  
**YOU HAVE ALL 6 PAGES OF THE EXAM BEFORE BEGINNING**  
**(not including the cover sheet).**  
**GOOD LUCK!**

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1. Convert the following to the given units: **(15 Points)**

(a) 200 W/(m<sup>2</sup> K) to Btu/(hr ft<sup>2</sup> °F) (5 points)

(b) 20.75 (psia)(ft<sup>3</sup>)/(lb-mol °R) to (L)(atm)/(gmol K) (5 points)

(c) A bucket contains 10 lb of water. If the specific heat ( $C_p$ ) of H<sub>2</sub>O is 4.17 kJ/(kg °C), what is its enthalpy change ( $\Delta H$ ) if the temperature is increased from 40°C to 90°C. Give the answer in Btu. (5 points)

$$\Delta H = mC_p (T_2 - T_1)$$

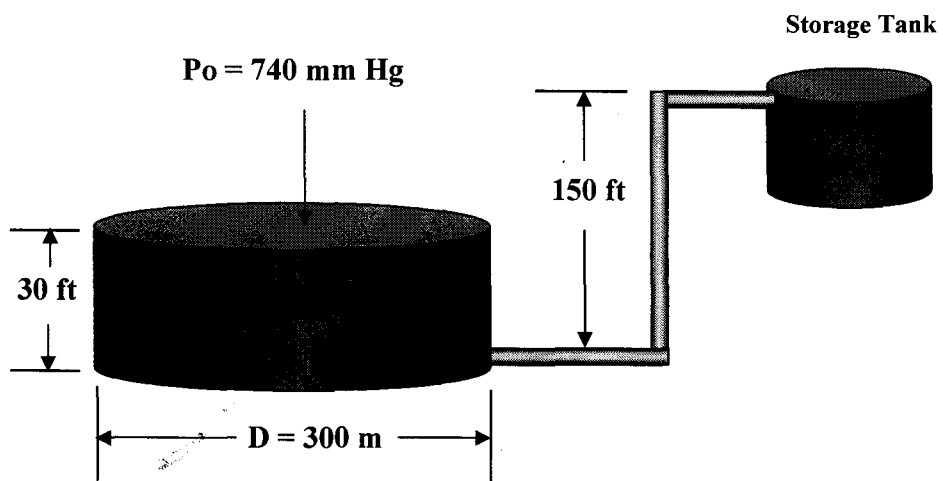
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<b>Constants:</b>	$g = 32.2 \text{ ft/s}^2 = 9.81 \text{ m/s}^2$	$1 \text{ lb}_m = 0.454 \text{ kg}$
	$g_c = 32.174 \text{ ft-lb}_m / (\text{lb}_f\text{-s}^2)$	$1 \text{ ft} = 0.3048 \text{ m}$
	$1 \text{ cp} = 1 \times 10^{-2} \text{ g}/(\text{cm-s})$	$1 \text{ m}^3 = 264.172 \text{ gal}$
	$1 \text{ psia} = 1 \text{ lb}_f/\text{in}^2 = 6.89476 \text{ kPa}$	$1 \text{ Pa} = 1 \text{ N/m}^2 = 1 \text{ kg}/(\text{m-s}^2)$
	$1 \text{ K} = 1.8^\circ\text{R}$	$1^\circ\text{C} = 1.8^\circ\text{F}$
	$\rho_{\text{H}_2\text{O}} = 62.4 \text{ lb}_m/\text{ft}^3 = 1 \text{ g/cm}^3$	$1 \text{ J/s} = 1 \text{ W (Watt)}$

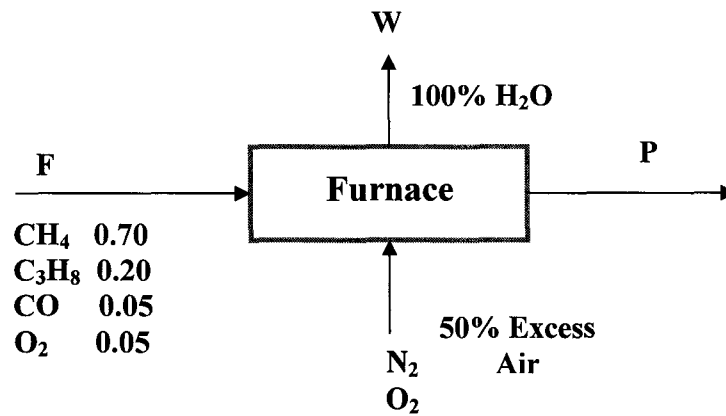
**Equations:** Pressure = Force/Area  
Static Pressure:  $P = \rho gh + P_o$

2. A centrifugal pump is to be used to pump water from the bottom of the PSU reservoir to a storage tank 150 feet above the surface of the reservoir. The pumping rate is 0.4 gal/min, and the water temperature is 20°C. The diameter of the reservoir is 300 m. (Neglect pipe friction kinetic energy effects, or factors involving the pump efficiency.) (20 Points)

- (a) What is the pressure in psig that the pump must develop in order to deliver a flow rate of 0.4 gal/min of water to the storage tank? (10 points)
- (b) What is the total force exerted on the bottom of reservoir in Newton? (10 points)  
(Hint: Determine the total pressure at the bottom of the reservoir in Pa units.)

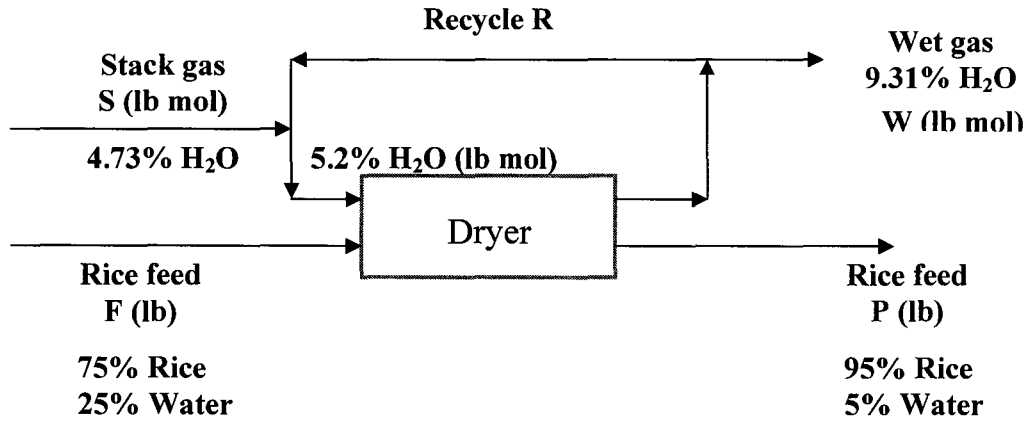


3. A mixture of 70% CH<sub>4</sub>, 20% C<sub>3</sub>H<sub>8</sub>, 5% CO and 5% O<sub>2</sub> is burned in a furnace with 50% excess air. If no CO leaves the furnace, determine the following information: (20 points)

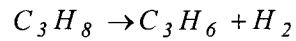


- The moles of air entering the furnace (10 points)
- The moles of water produced (5 points)
- The Orsat Analysis of the flue gas (5 points)

4. To save energy, stack gas from a furnace is used to dry rice. The flow sheet and known data are shown below. What is the amount of recycle gas (in lb mol) per 100 lb of P if the concentration of water in the gas stream entering the dryer is 5.20%? (Hint: Do a mass balance for Rice and Water. Then, do mole balances for water and gas.) (20 points)



5. The process shown in the figure below is the dehydrogenation of propane ( $C_3H_8$ ) to propylene ( $C_3H_6$ ) according to the reaction.



The conversion of propane to propylene based on the total propane feed into the reactor at  $F_2$  is 60%. The product flow rate  $F_5$  is 50 kg mol/hr. Calculate all the six flow rates  $F_1$  to  $F_6$  in kg mol/hr. (25 points)

