

PRINCE OF SONGKLA UNIVERSITY
FACULTY OF ENGINEERING

Midterm Examination : Semester I

Academic Year : 2005

Date : 7 August 2005

Time : 09.00 — 12.00

Subject : 230 — 531 Membrane Separation Processes

Room : R 300

Student Name: ID no. :

Number of questions : 4

Time : 3 hours

Total marks : 100

Calculators are allowed.

Books and notes are allowed.

Question	Full Marks	Marks Received
1	30	
2	25	
3	20	
4	25	
Total	100	

ทูลจริตในการสอบโทษขันต่ำคือ ปรับตกในรายวิชาที่ทูลจริต และพัทการเรียน 1 ภาคการศึกษา

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- 1) a) What is a composite membrane? Explain with help of a sketch of the membrane microscopic structure.
(5 marks)
- b) Please describe how to construct a spiral-wound module. Explain the flow direction of all streams through this type of membrane module.
(5 marks)
- c) What are the advantages and disadvantages of ceramic membranes in comparison to polymeric membranes?
(5 marks)
- d) Provide a sketch of a ceramic membrane module. Show stream flow paths in ultrafiltration. Give some details on pore sizes and materials of construction.
(5 marks)
- e) Briefly explain transport mechanisms of pervaporation for a silicone membrane.
(5 marks)
- f) Fuel cell vehicles are expected to offer an extremely quiet ride, improved fuel economy and zero emissions. Briefly describe the basic principle of fuel cell operation.
(5 marks)

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2) It is desired to determine the membrane area needed to separate an air stream using a membrane thickness $t = 2.54 \times 10^{-3}$ cm. with an oxygen permeability of $P_A'' = 500 \times 10^{-10}$ cm³ (STP).cm/(s.cm².cm Hg). An $\alpha^* = 10$ for oxygen permeability divided by nitrogen permeability will be used. The feed rate is $L_f = 1 \times 10^6$ cm³ (STP)/s , $x_f = 0.209$ and the fraction cut $\theta = 0.35$. The pressures selected for use are $p_h = 190$ cm Hg and $p_l = 19$ cm Hg. Assuming the crossflow model, calculate permeate composition, the reject composition, and the membrane area.

Simulation results at some values of θ or θ^* are given in the table below.

Use the graph provided on page 8 for determination of membrane area..

Perform only one and first trial using $x=0.05$.

θ or θ^*	x	y_p	F_i
0	0.209	0.6550	0.6404
0.1482	0.1420	0.5940	0.9603
0.2000	0.1190	0.5690	1.1520

(25 marks)

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Answer to Q2.(continued)

Graph for determination of membrane area in Q2

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- 3) The following experimental data were obtained from the pervaporation of liquid mixtures of ethanol (1) and water (2) at feed temperature of 60°C for a permeate pressure of 76 mmHg, using a commercial silicone membrane.

wt % ethanol		Total Permeation Flux Kg/m ² -h
Feed	Permeate	
60.0	90.0	3.2

Molecular weights for ethanol and water are 46.07 and 18.02 respectively.

At 60°C, vapour pressures for ethanol and water are 352 and 149 mmHg, respectively.

Liquid-phase activity coefficients at 60°C for the feed ethanol(1)-water(2) system are given by the van Laar equations as:

$$\ln \gamma_1 = 1.6276 \left[\frac{0.9232x_2}{1.6276x_1 + 0.9232x_2} \right]^2$$

$$\ln \gamma_2 = 0.9232 \left[\frac{1.6276x_1}{1.6276x_1 + 0.9232x_2} \right]^2$$

- a) Calculate values of permeance for ethanol and water in kmol/(h-m²-mmHg).
 b) The separation factors for ethanol $\alpha_{1,2}$.

(20 marks)

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- 4) A reverse osmosis plant is being used to treat $85 \text{ m}^3/\text{h}$ feed of brackish water at 27°C containing $2,000 \text{ mg/l}$ NaCl to produce potable water with 200 mg/l of dissolved NaCl. The plant uses cellulose acetate membranes arranged in two stages. The first stage consists of 10 pressure vessels and one element per vessel. The second stage consists of 8 pressure vessels and one element per vessel. The feed-side pressure at the first stage is 7 atm, while the concentrate from the first stage is fed at 16 atm to the second stage. The recovery for the first and second stages are 30% and 10%, respectively. The permeability constant for water $A_B = 4.20 \times 10^{-4} \text{ kg /s.m}^2.\text{atm}$ and permeability constant for NaCl solute $A_A = 4.0 \times 10^{-7} \text{ m/s}$. Molecular weight of NaCl is 58.5 g/mol . $R = 82.057 \times 10^{-3} \text{ m}^3.\text{atm}/(\text{k mol.K})$.

Calculate the followings:

- The overall membrane area for the first stage in m^2 .
- The salt passage in the first stage in $\text{kg}/(\text{m}^2.\text{s})$
- The total permeate product flow in m^3/h .

(25 marks)

----- End of questions.