Name:	Student ID	
Nickname:	Group:	

Prince of Songkla University Faculty of Engineering

Exam: Mid-Term, Semester I

Academic Year: 2009 – 2010

Date: July 27, 2009

Time: 1:30 – 4:30 PM

Subject: 230-301

Room: A203

Basic Chemical Engineering I

ทุจริตในการสอบโทษขั้นต่ำคือ ปรับตกในรายวิชาที่ทุจริต และพักการเรียน 1 ภาคการศึกษา

Instructions: There are a total of 5 problems and 9 pages (not including this page). Place your name and the student ID number on every page. Students are allowed to use <u>only</u> a pen or pencil and a calculator. They can also bring in 1 sheet of A4 front side only, a Conversions Table, and a Dictionary. No exams are allowed to leave the room.

Points Distribution (For Grader Only)		
Problem	Points Value	Score
1	25	711
2	20	
3	15	
4	20	
5	20	
Total	100	

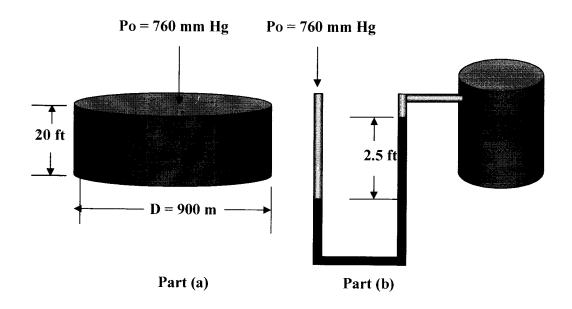
Exam prepared by Ram Yamsaengsung July 22, 2009

PLEASE CHECK TO MAKE SURE THAT
YOU HAVE ALL 9 PAGES OF THE EXAM BEFORE BEGINNING
(not including the cover sheet).
GOOD LUCK!

Prince of Songkla University Faculty of Engineering

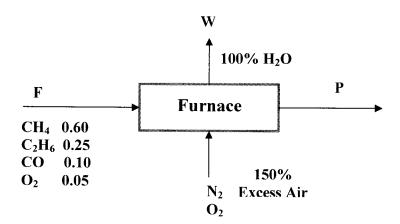
Exam: Mid-Term, Semester I Date: July 27, 2009 Subject: 230-301 Basic Chemical Engineering I	Academic Year: 2009 – 2010 Time: 1:30 – 4:30 PM Room: A203
1. Conversions of Units: (25 Points)	
1.1 Convert	
(a) $0.55 \text{ g/(min)(in}^3)$ to $lb_m/(hr)(ft^3)$	(5 points)
(b) 1.35 Btu/[(hr)(ft ²)(°F/ft)] to kJ/[((day)(m ²)(°C/cm)] (5 points)
1.2 Water is flowing through a 2.5-inch diam	meter pipe with a velocity of 3 m/s.
(a) What is the kinetic energy of the	water in (ft)(lb _f)/lb? (5 points)
(b) What is the flow rate in gal/min?	(10 points)

- 2. From the figures below, answer the following questions. (20 Points)
 - (a) What is the total force exerted on the bottom of reservoir in Newton? (10 points) (Hint: Determine the total pressure at the bottom of the reservoir in Pa units.)
 - (b) What is the pressure inside the storage tank in psig if water is used as the fluid inside the manometer? (10 points)



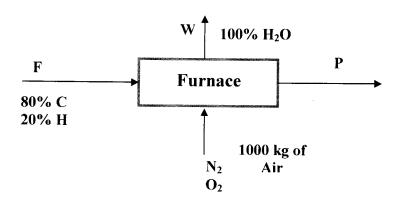
3.	To prepare a solution of 65% sulfuric acid, a dilute waste acid containing 35% H ₂ SO ₄ is fortified with a purchased acid containing 95% H ₂ SO ₄ . How many kilograms of the purchased acid must be bought for each 100 kg of dilute acid? (15 Points)		

4. A mixture of 60% CH₄, 25% C₂H₆, 10% CO and 5% O₂ is burned in a furnace with 200% excess air. If no CO, CH₄, and C₂H₆ leave the furnace, determine the following information: **(20 points)**



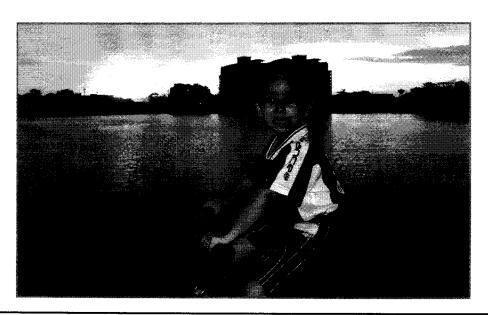
- (a) The moles of air entering the furnace (10 points)
- (b) The moles of water produced (5 points)
- (c) The Orsat Analysis of the flue gas (5 points)

5. Fifty kilograms of coal (analysis 80% C and 20% H ignoring the ash) are burned with 1000 kg of air, yielding a gas having an Orsat analysis in which the ratio of CO_2 to CO is 3 to 2. **(20 points)**



- (a) What is the percent excess air? (10 points)
- (b) What is the Orsat analysis of the flue gas? (10 points)

BONUS: What is the name of the restaurant that N'Bright is sitting in below? (5 Points)





CONGRATULATIONS! END OF EXAM!





Constants:

 $g = 32.2 \text{ ft/s}^2 = 9.81 \text{ m/s}^2$

 $g_c = 32.174 \text{ ft-lb}_m / (\text{lb}_{\text{f}} - \text{s}^2)$ 1 Btu = 1.055 x 10³ J

1 psia = $1 lb_f/in^2 = 6.89476 kPa$

 $1\Delta K = 1.8\Delta^{\circ}R$

 $\rho_{\rm H2O} = 62.4 \ \rm lb_m/ft^3 = 1 \ g/cm^3$

 $11b_{m} = 0.454 \text{ kg}$ 1 ft = 0.3048 m

 $1 \text{ m}^3 = 264.172 \text{ gal}$

 $1 \text{ Pa} = 1 \text{ N/m}^2 = 1 \text{ kg/(m-s}^2)$

 $1\Delta^{\circ}C = 1.8\Delta^{\circ}F$ 1 J/s = 1 W (Watt)

Equations: Pressure = Force/Area

Static Pressure: $P = \rho gh + Po$

Area of Circle = $\pi D^2/4$