

Name: \_\_\_\_\_ Student ID \_\_\_\_\_

**Prince of Songkla University  
Faculty of Engineering**

**Final Exam: Semester I  
Date: September 29, 2009  
Subject: 230-301 – Basic Chemical Engineering I**

**Academic Year: 2009 – 10  
Time: 1:30 – 4:30 PM  
Room: R200**

---

**Part I: Energy Balance**

**Instructions: There are a total of 2 problems. You have 60 minutes to work on them. The points for each problem are not distributed evenly. Place your name and the student ID number on every page. This is a CLOSED BOOK exam. Students are allowed to use only a pen or pencil, a calculator, and a pocket or talking dictionary. No notes are allowed in the exam. No exams are allowed to leave the room.**

<b>Points Distribution (For Grader Only)</b>		
<b>Problem</b>	<b>Points Value</b>	<b>Score</b>
<b>1</b>	<b>25</b>	
<b>2</b>	<b>25</b>	
<b>Total</b>	<b>50</b>	

**Exam prepared by  
Ram Yamsaengsung  
September 22, 2009**

**PLEASE CHECK TO MAKE SURE THAT  
YOU HAVE 6 PAGES OF THE EXAM AND TWO SHEETS FROM THE  
STEAM TABLE BEFORE BEGINNING**

**GOOD LUCK!!!**

### Final Exam Part I

1. Ammonia is burned in air with a resulting flue gas having an Orsat analysis of 6.5% O<sub>2</sub> and 93.5% N<sub>2</sub>. Determine the following if the NH<sub>3</sub> flow rate is 28.7 m<sup>3</sup>/min at 45°C and 125.6 kPa. **(25 points)**
- The m<sup>3</sup> of air required at 25°C and 100.0 kPa used per m<sup>3</sup> of ammonia used. **(5 points)**
  - The percent excess air used. **(5 points)**
  - The m<sup>3</sup> of actual air at 25°C and 100.0 kPa. **(5 points)**
  - The amount of heat that must be removed (kJ) in order to cool the water vapor produced from 100°C to liquid water at 20°C. Assume the exiting water vapor is at 101.3 kPa. **(10 points)**

**Conversions:** 1Btu = 1.055 kJ      1 lb = 0.454 kg  
1 atm = 14.696 psia

**Constants:** R = 82.06 (cm<sup>3</sup>-atm/g mol-K) = 8.314 (kPa-m<sup>3</sup>/kg mol-K)

**Equations:**  $PV = nRT$   

$$P_i = \left( \frac{n_i}{n_{tot}} \right) P_{tot}$$

$$\Delta E = Q + W - \Delta[(H + K + P)]$$

$$\Delta H = n \int_{T_1}^{T_2} C_p dT = n(h_2 - h_1)$$

$$\Delta \hat{H}_{rxn}^0 = \sum \Delta \hat{H}_f^0 \text{ products} - \sum \Delta \hat{H}_f^0 \text{ reactants}$$

$$\Delta H_i = n_i \left[ \int_{25}^{T_i} C_{p,i} dT + \Delta \hat{H}_{fi}^0 + \Delta \hat{H}_{\text{phase change}} \right]$$

2. In a pasteurization process, steam is used to raise the temperature of the orange juice from 25°C to 73°C. If 2500 kg/hr of orange juice is fed into the heat exchanger, determine the following: **(25 Points Total)**

- (a) Determine the amount of heat gained by the orange juice in kJ/hr. **(10 points)**
- (b) Determine the amount of saturated steam (in **kg/hr**) required if the steam enters as saturated vapor at 1 atm and leaves as 25% saturated liquid. **(15 points)**

The  $C_p$  of the orange juice is 4.00 kJ/kg °C.

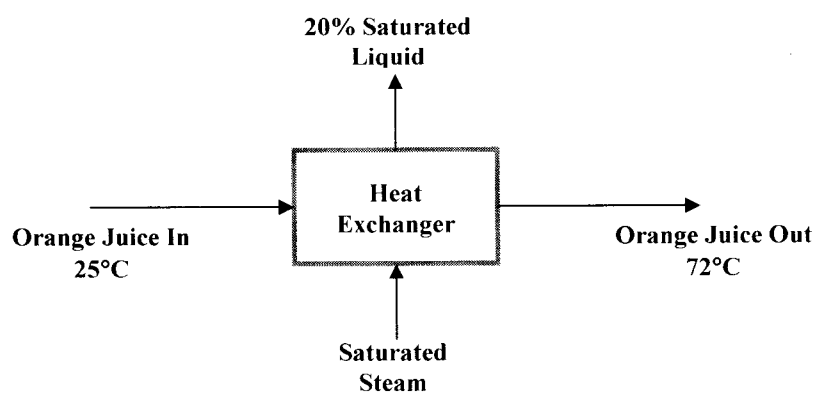


Table 3. Superheated Steam - Continued

Abs Press. (Lb/Sq In.)	Temp (°F)	Sat. Steam		Water		Sat. Steam		Water		Sat. Steam		Water		Sat. Steam		Water	
		$v$	$h$	$v$	$h$	$v$	$h$	$v$	$h$	$v$	$h$	$v$	$h$	$v$	$h$	$v$	$h$
1100	1500	0.0245	0.0267	0.0296	0.0335	0.0386	0.0443	0.0503	0.0562	0.0620	0.0676	0.0727	0.0772	0.0818	0.0868	0.0922	0.0976
1150	1500	0.0243	0.0263	0.0290	0.0325	0.0370	0.0423	0.0478	0.0534	0.0588	0.0641	0.0691	0.0739	0.0787	0.0837	0.0892	0.0946
1200	1500	0.0241	0.0260	0.0284	0.0317	0.0357	0.0405	0.0456	0.0508	0.0560	0.0610	0.0659	0.0704	0.0751	0.0800	0.0850	0.0900
1250	1500	0.0238	0.0256	0.0279	0.0309	0.0346	0.0390	0.0437	0.0486	0.0535	0.0583	0.0629	0.0673	0.0719	0.0766	0.0814	0.0862
1300	1500	0.0236	0.0253	0.0275	0.0302	0.0336	0.0376	0.0420	0.0466	0.0512	0.0558	0.0602	0.0645	0.0689	0.0734	0.0780	0.0826
1350	1500	0.0235	0.0251	0.0271	0.0297	0.0328	0.0364	0.0405	0.0448	0.0492	0.0535	0.0577	0.0619	0.0661	0.0704	0.0748	0.0792
1400	1500	0.0233	0.0248	0.0267	0.0291	0.0320	0.0354	0.0392	0.0432	0.0474	0.0515	0.0555	0.0595	0.0634	0.0673	0.0712	0.0751
1450	1500	0.0231	0.0246	0.0264	0.0287	0.0314	0.0345	0.0380	0.0418	0.0458	0.0496	0.0533	0.0570	0.0606	0.0643	0.0680	0.0717
1500	1500	0.0228	0.0244	0.0261	0.0282	0.0308	0.0337	0.0369	0.0405	0.0440	0.0475	0.0509	0.0543	0.0577	0.0610	0.0644	0.0677
1550	1500	0.0228	0.0242	0.0258	0.0278	0.0302	0.0329	0.0360	0.0393	0.0429	0.0464	0.0498	0.0531	0.0564	0.0597	0.0630	0.0663

SYSTÈME INTERNATIONAL (SI METRIC) UNITS

Table 1. Saturated Steam and Saturated Water: Temperature Table

Temp °C	Temp °F	Pressure kPa		Volume, m <sup>3</sup> /kg		Enthalpy, kJ/kg		Entropy, kJ/kg·K		Temp °C
		h <sub>g</sub>	h <sub>f</sub>	v <sub>g</sub>	v <sub>f</sub>	h <sub>g</sub>	h <sub>f</sub>	s <sub>g</sub>	s <sub>f</sub>	
0	32	0.6108	0.0010002	206.31	-0.04	2501.6	2501.6	-0.0002	0.1579	0
5	41	0.8718	0.0010000	206.16	0.00	2501.6	2501.6	0.0000	0.1575	5
10	50	1.2270	0.0010003	106.43	41.99	2477.9	2518.9	0.1510	0.8020	10
15	59	1.7040	0.0010008	77.96	62.94	2466.1	2529.0	0.2243	0.8785	15
20	68	2.377	0.0010017	57.84	83.86	2454.3	2538.2	0.2983	0.9371	20
25	77	3.166	0.0010029	43.40	104.77	2442.5	2547.3	0.3670	0.9822	25
30	86	4.241	0.0010043	32.83	125.66	2430.7	2556.4	0.4365	1.0181	30
35	95	5.622	0.0010060	25.25	146.58	2418.8	2565.4	0.5049	1.0454	35
40	104	7.375	0.0010078	19.546	167.45	2406.9	2574.4	0.5721	1.0661	40
45	113	9.582	0.0010098	15.276	188.35	2394.9	2583.3	0.6383	1.0816	45
50	122	12.335	0.0010121	12.046	209.28	2382.9	2592.2	0.7035	1.0926	50
55	131	15.741	0.0010145	9.579	230.17	2370.8	2601.0	0.7677	1.1000	55
60	140	19.920	0.0010171	7.679	251.09	2358.6	2609.7	0.8310	1.1043	60
65	149	25.010	0.0010199	6.202	272.03	2346.3	2618.3	0.8933	1.1068	65
70	158	31.16	0.0010228	5.046	292.97	2334.0	2626.9	0.9548	1.1075	70
75	167	38.55	0.0010259	4.134	313.93	2321.5	2635.4	1.0154	1.1068	75
80	176	47.36	0.0010292	3.409	334.92	2308.8	2643.8	1.0753	1.1043	80
85	185	57.80	0.0010326	2.829	355.91	2296.1	2652.0	1.1343	1.1000	85
90	194	70.11	0.0010361	2.3613	376.94	2283.2	2660.1	1.1925	1.0925	90
95	199	84.53	0.0010398	1.9822	397.99	2270.2	2668.2	1.2501	1.0816	95
100	212	101.33	0.0010437	1.6730	419.06	2256.9	2676.0	1.3069	1.0661	100
105	217	120.80	0.0010477	1.4193	440.17	2243.6	2683.7	1.3630	1.0454	105
110	228	143.27	0.0010519	1.2099	461.32	2230.0	2691.3	1.4185	1.0200	110
115	239	169.06	0.0010562	1.0363	482.50	2216.2	2698.7	1.4733	0.9925	115
120	250	198.54	0.0010606	0.8915	503.72	2202.2	2706.0	1.5276	0.9601	120
125	259	232.1	0.0010652	0.7702	524.99	2188.0	2713.0	1.5813	0.9238	125
130	270	270.1	0.0010700	0.6681	546.31	2173.6	2718.9	1.6344	0.8844	130
135	281	313.1	0.0010750	0.5818	567.68	2158.9	2726.6	1.6869	0.8418	135
140	292	361.4	0.0010801	0.5085	589.10	2144.0	2733.1	1.7390	0.7968	140
145	303	415.5	0.0010853	0.4460	610.59	2128.7	2739.3	1.7908	0.7508	145
150	314	476.0	0.0010908	0.3924	632.15	2113.2	2745.4	1.8416	0.7038	150
155	325	543.3	0.0010964	0.3464	653.77	2097.4	2751.2	1.8923	0.6561	155
160	336	618.1	0.0011022	0.3068	675.47	2081.3	2756.7	1.9425	0.6078	160
165	347	700.8	0.0011082	0.2724	697.25	2064.8	2762.0	1.9923	0.5594	165
170	358	792.0	0.0011145	0.2426	719.12	2047.9	2767.1	2.0410	0.5114	170
175	369	892.4	0.0011209	0.2165	741.07	2030.7	2771.8	2.0896	0.4641	175
180	380	1002.7	0.0011275	0.1938	763.12	2013.2	2776.3	2.1383	0.4176	180
185	391	1123.3	0.0011344	0.1738	785.26	1995.2	2780.4	2.1876	0.3718	185
190	402	1255.1	0.0011415	0.1563	807.52	1976.7	2784.3	2.2356	0.3274	190

LE C.1 (cont.)

Absolute pressure			Specific volume				Enthalpy	
Lb/in. <sup>2</sup> <i>p</i>	In. Hg 32°F	Sat. Liquid <i>v<sub>f</sub></i>	Evap. <i>v<sub>fg</sub></i>	Sat. Vapor <i>v<sub>g</sub></i>	Sat. Liquid <i>h<sub>f</sub></i>	Evap. <i>h<sub>fg</sub></i>	Sat. Vapor <i>h<sub>g</sub></i>	
3.198	6.511	0.01631	111.88	111.90	111.88	1011.3	1123.2	
3.363	6.847	0.01632	106.72	106.74	113.88	1010.2	1124.1	
3.536	7.199	0.01633	101.82	101.84	115.87	1009.0	1124.9	
3.716	7.566	0.01634	97.18	97.20	117.87	1007.8	1125.7	
3.904	7.948	0.01635	92.79	92.81	119.87	1006.7	1126.6	
4.100	8.348	0.01636	88.62	88.64	121.87	1005.5	1127.4	
4.305	8.765	0.01637	84.66	84.68	123.87	1004.4	1128.3	
4.518	9.199	0.01638	80.90	80.92	125.87	1003.2	1129.1	
4.739	9.649	0.01639	77.37	77.39	127.87	1002.0	1129.9	
4.970	10.12	0.01640	74.00	74.02	129.88	1000.8	1130.7	
5.210	10.61	0.01642	70.79	70.81	131.88	999.7	1131.6	
5.460	11.12	0.01643	67.76	67.78	133.88	998.5	1132.4	
5.720	11.65	0.01644	64.87	64.89	135.88	997.3	1133.2	
5.990	12.20	0.01645	62.12	62.14	137.89	996.1	1134.0	
6.272	12.77	0.01646	59.50	59.52	139.89	995.0	1134.9	
6.565	13.37	0.01647	57.01	57.03	141.89	993.8	1135.7	
6.869	13.99	0.01648	54.64	54.66	143.90	992.6	1136.5	
7.184	14.63	0.01650	52.39	52.41	145.90	991.4	1137.3	
7.510	15.29	0.01651	50.26	50.28	147.91	990.2	1138.1	
7.849	15.98	0.01652	48.22	48.24	149.92	989.0	1138.9	
8.201	16.70	0.01653	46.28	46.30	151.92	987.8	1139.7	
8.566	17.44	0.01654	44.43	44.45	153.93	986.6	1140.5	
8.944	18.21	0.01656	42.67	42.69	155.94	985.3	1141.3	
9.336	19.01	0.01657	40.99	41.01	157.95	984.1	1142.1	
9.744	19.84	0.01658	39.38	39.40	159.95	982.8	1142.8	
10.168	20.70	0.01659	37.84	37.86	161.96	981.5	1143.5	
10.605	21.59	0.01662	36.38	36.40	163.97	980.3	1144.3	
11.057	22.51	0.01663	34.98	35.00	165.98	979.0	1145.0	
11.525	23.46	0.01665	33.65	33.67	167.99	977.8	1145.8	
12.010	24.45	0.01666	32.37	32.39	170.01	976.6	1146.6	
12.512	25.47	0.01666	31.15	31.17	172.02	975.3	1147.3	
13.031	26.53	0.01667	29.99	30.01	174.03	974.1	1148.1	
13.568	27.62	0.01669	28.88	28.90	176.04	972.8	1148.8	
14.123	28.75	0.01670	27.81	27.83	178.06	971.5	1149.6	
14.696	29.92	0.01672	26.81	26.83	180.07	970.3	1150.4	
15.291		0.01674	25.35	25.37	186.10	968.3	1151.4	
17.188		0.01677	23.14	23.16	188.14	965.2	1153.3	
18.915		0.01681	21.15	21.17	193.18	961.9	1155.1	

*v* = specific volume, ft<sup>3</sup>/lb. *h* = enthalpy, Btu/lb.

TABLE C.1 (cont.)

Temp. Fahr. <i>t</i>	Absolute pressure		Specific volume				Enthalpy	
	Lb/in. <sup>2</sup> <i>p</i>	In. Hg 32°F	Sat. Liquid <i>v<sub>f</sub></i>	Evap. <i>v<sub>fg</sub></i>	Sat. Vapor <i>v<sub>g</sub></i>	Sat. Liquid <i>h<sub>f</sub></i>	Evap. <i>h<sub>fg</sub></i>	Sat. Vapor <i>h<sub>g</sub></i>
230	20.78		0.01684	19.371	19.388	198.22	958.7	1156.9
235	22.80		0.01688	17.761	17.778	203.28	955.3	1158.6
240	24.97		0.01692	16.307	16.324	208.34	952.1	1160.4
245	27.31		0.01696	15.010	15.027	213.41	948.7	1162.1
250	29.82		0.01700	13.824	13.841	218.48	945.3	1163.8
255	32.53		0.01704	12.735	12.752	223.56	942.0	1165.6
260	35.43		0.01708	11.754	11.771	228.65	938.6	1167.3
265	38.54		0.01713	10.861	10.878	233.74	935.3	1169.0
270	41.85		0.01717	10.053	10.070	238.84	931.8	1170.6
275	45.40		0.01721	9.313	9.330	243.94	928.2	1172.1
280	49.20		0.01726	8.634	8.651	249.06	924.6	1173.7
285	53.25		0.01731	8.015	8.032	254.18	921.0	1175.2
290	57.55		0.01735	7.448	7.465	259.31	917.4	1176.7
295	62.13		0.01740	6.931	6.948	264.45	913.7	1178.2
300	67.01		0.01745	6.454	6.471	269.60	910.1	1179.7
305	72.18		0.01750	6.014	6.032	274.76	906.3	1181.1
310	77.68		0.01755	5.610	5.628	279.92	902.6	1182.5
315	83.50		0.01760	5.239	5.257	285.10	898.8	1183.9
320	89.65		0.01765	4.897	4.915	290.29	895.0	1185.3
325	96.16		0.01771	4.583	4.601	295.49	891.1	1186.6
330	103.03		0.01776	4.292	4.310	300.69	887.1	1187.8
335	110.31		0.01782	4.021	4.039	305.91	883.2	1189.1
340	117.99		0.01788	3.771	3.789	311.14	879.2	1190.3
345	126.10		0.01793	3.539	3.557	316.38	875.1	1191.5
350	134.62		0.01799	3.324	3.342	321.64	871.0	1192.6
355	143.58		0.01805	3.126	3.144	326.91	866.8	1193.7
360	153.01		0.01811	2.940	2.958	332.19	862.5	1194.7
365	162.93		0.01817	2.768	2.786	337.48	858.2	1195.7
370	173.33		0.01823	2.607	2.625	342.79	853.8	1196.6
375	184.23		0.01830	2.458	2.476	348.11	849.4	1197.5
380	195.70		0.01836	2.318	2.336	353.45	844.9	1198.4
385	207.71		0.01843	2.189	2.207	358.80	840.4	1199.2
390	220.29		0.01850	2.064	2.083	364.17	835.7	1199.9
395	233.47		0.01857	1.9512	1.9698	369.56	831.0	1200.6
400	247.25		0.01864	1.8446	1.8632	374.97	826.2	1201.2
405	261.67		0.01871	1.7445	1.7632	380.40	821.4	1201.8
410	276.72		0.01878	1.6508	1.6696	385.83	816.6	1202.4