

**PRINCE OF SONGKLA UNIVERSITY
FACULTY OF ENGINEERING**

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Name Student ID

อนุญาตให้นำเอกสารและเครื่องคำนวณทุกชนิดเข้าห้องสอบ
ทุจริตในการสอบโทษขั้นต่ำคือปรับตกในรายวิชาที่ทุจริต
และพักการศึกษา 1 ภาคการศึกษา

Question #	Total Score	Score
1	10	
2	35	
3	20	
4	30	
5	20	
6	25	
Total	140	

ดร. พรศิริ แก้วประดิษฐ์
ผู้ออกข้อสอบ

1. (10 points), These sentences are ✓ “True” or ✗ “False”

- 1.1. Flux is a transfer rate per unit volume.
- 1.2. Driving force of mass transfer is concentration gradient.
- 1.3. In film theory, it is assumed that eddy diffusivity is zero within the effective film thickness.
- 1.4. A rate of transfer is reduced if boiling point elevation (BPE) of the solution increases.
- 1.5. BPE of the solution depends only on the concentration of solute.
- 1.6. Temperature drop in the evaporator is the difference between the superheated temperature of the steam and the temperature in vapor space.
- 1.7. For dilute solution, heat of dilution and BPE can be negligible.
- 1.8. To concentrating β - carotene, residence time in a evaporator should be minimized.
- 1.9. In two – phase theory, if $\frac{k_x}{mk_y} \ll 1$ it means that the mass – transport resistance of the gas phase has large effect.
- 1.10. Turbulent diffusion provides higher transfer rate due to addition of eddy diffusivity term.

2. (35 points) A 20% NaOH solutions is to be concentrated to 65% in a single – effect evaporator with a vertical tube 6 m in diameter and 15 m long. The feed rate is 60,000 kg/h at 40.56°C. The boiling point of water at the absolute pressure in vapor space is 110°C. Steam is available at a gauge pressure of 261.8 lb_f/in².
- 2.1. (20 points), calculate the steam requirement in kg/hr
- 2.2. (5 points), calculate the economy
- 2.3. (10 points), estimate the overall heat – transfer coefficient in W/m². °C

3. (20 points) Carbon tetrachloride (CCl_4) evaporates into a tube 1.02 in diameter containing oxygen. The distance between the CCl_4 liquid level and the top of the tube is 17.1 cm. The total pressure on the system is 755 mmHg, and the temperature is 0°C . The vapor pressure of CCl_4 at that temperature is 33 mmHg and its density is 1.629 g/cm^3 .
- 3.1. (10 points), calculate the molar flux of CCl_4 , if it is found that 0.0208 cm^3 of CCl_4 evaporate in a 10 hours period.
- 3.2. (10 points), estimate the diffusivity of the gas pair $\text{O}_2 - \text{CCl}_4$ in cm^2/s

4. (30 points) A spherical drop of water 0.05 cm in diameter is falling at a velocity 215 cm/s through dry, still air at 105°F and 1 atm. The vapor pressure of water at 105°F is 0.0247 atm.
- 4.1. (15 points), estimate the mass transfer coefficient, k_c
- 4.2. (5 points), estimate the mass transfer coefficient relating partial pressure, k_G in mol/cm².s.atm
- 4.3. (10 points), calculate the evaporation rate

5. (20 points) A tube 0.2 cm in diameter is filled with liquid n-heptane at 21°C. The diffusivity of n-heptane in air at that temperature is 0.071 cm²/s and the vapor pressure is 0.05 atm.
- 5.1. (10 points), estimate the mass transfer coefficient relating mole – fraction in gas phase, k_y if the liquid level is decreased 1 cm from the top
- 5.2. (10 points), calculate the rate of decrease of the liquid level in cm/h (molecular weight and density of n-heptanes is 100.2 g/mol, 0.66 g/cm³)

6. (25 points) A dilute solution of organic colloids in water is to be concentrated from 8 to 45 % solids in a forced – circulation evaporator. The steam is available at 249°F, and a pressure in a vapor space is maintained at 102 mmHg. The feed rate to the evaporator is 20,000 kg/h, and the specific heat of the feed solution is 3.77 J/g.°C.
- 6.1. (5 points), determine the capacity of the evaporator
- 6.2. (20 points), calculate the feed temperature and the heating load, if the economy is required at 1.